Laboratory Experiment Proposal Submission

Experimental Details

Experiment location: E134

Experiment title: Interaction of tubulin with tethered lipid membranes

Experiment date: 7/23/2013

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Chemicals Used

Chemical Name

Sodium Chloride Potassium Chloride HEPES Buffer

Sodium PIPES

Ethanol Chloroform

Sodium Phosphate

EGTA (ethylene Glycol Tetraacetic Acid)

EDTA (ethylene Diamine Tetraacetic Acid)

Sulfuric Acid

Hydrogen Peroxide

<u>Health</u>	<u>Flammability</u>	<u>Reactivity</u>
0	0	0
2	0	0
2 2 2 2 2	1	0
2	1	0
2	3	0
2	0	0
0	0	0
0	1	0
2	0	0
2 3 3	0	2
3	0	1

Special Hazards	3
NONE	
NONE	
COR	
OX	

Reactants and Resulting Samples

Chemical Name

Gold-coated Silicon Wafer With Self-assembled

Monolayer

<u>Hazardous?</u> N

Known Hazards

Required Safety Equipment

- ✓ Glove Chemical
- Glove Cold
- Glove Neoprene
- Safety Glasses
- Hood Acids
- Hood Organics

Required Laboratory Equipment

- Balance
- Hot Plate
- Ph Meter
- Lyophilizer
- Ultrasonic Bath
- Vacuum Pump

Experimental	Write I	Up
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Optimal experimental procedure has not yet been determined, because it is not known what will work. The basic idea is as follows:

- 1. Lipids (usually precipitated from ~ 1 mL chloroform and lyophilized) are dissolved in the solution of interest by sonication. This solution will generally be a 1 M KCl/ 1 mM HEPES buffer at pH 7, but the pH could vary substantially (4-10) and the concentration as well (perhaps as low as 10 mM). Other buffers of interest are phosphate buffers with NaCl.
- 2. Bilayers are formed by exposing the self-assembled monolayer (SAM)-coated gold-on-silicon to the lipids.
- 3. Tubulin is received from the manufacturer as a powder. This is dissolved in a solution of PIPES/MgCl2/EGTA and stored in very small aliquots (~10-100 uL) of 50 uM solution. These aliquots are flash-frozen in liquid nitrogen (by lowering the tubes into a wide-mouth dewar containing a couple inches (< 1 liter) of liquid nitrogen and then storing them in a -80 C freezer until ready for use.
- 4. The tubulin solution will then be flowed across the bilayer surface after dilution to about 100 nM.
- 5. Characterization techniques to be used include neutron reflectivity and electrochemical impedance spectroscopy.
- 6. Piranha solution may occasionally be used to clean gold surfaces.
- 7. Self-assembled monolayers are formed from a very dilute ethanol solution of beta-mercaptoethanol and a lipid-like chemical made by Dave Vanderah at IBBR.
- 8. VDAC procedures are not yet determined.

Experimenter Signature:	Date:
Lab Responsible Signature:	Date: